(5+4+3)

# Reg. No.

## First Semester M.Sc. Degree Examination, December 2018 INDUSTRIAL CHEMISTRY **Environmental, Health and Safety Measures**

Time: 3 Hours

#### **Instructions**: 1) Answer **any five** guestions from Part – **A** and **any five** questions from Part – **B**.

2) Figures to the **right** indicate marks.

## PART – A

- 1. a) What is acid rain ? Mention its effects.
  - b) What is the use of chlorofluoro carbons? Why they have been banned?
  - c) A 50.0 g water sample was evaporated at 105 °C for 3 hours to give a concurrent value of 1.0 g. Calculate the total solids present in the sample.
  - d) A 100.0 mL water sample consumed 10.0 mL 0.02 N standard AgNO solution. Calculate the amount of chloride present in one liter of water sample.
  - e) What is sampling ? Give its significance.
  - f) Mention the chemical composition and application of tear gas.
  - g) Name any two personal protective equipments and their applications.
  - h) Write the emergency procedure to be followed in case of fire accident in the laboratory.

PART – B

#### Answer **any five** of the following :

- 2. a) What is photochemical smog? How it is formed and what are its effects on environment?
  - b) List the sources of SOx. Explain the principle and working of limestone injection process in the control of SOx emission.
  - c) Write a note on dispersion of air pollutants.

(5×2=10)

(5×12=60)

Max. Marks: 70

**ICS 404** 

#### **ICS 404**

- 3. a) What is green house effect ? Discuss the cause, consequence and control of green house effect.
  - b) Briefly explain the principle and application of wet scrubbers and electrostatic precipitators in the control of NOx emission. (6+6)
- 4. a) List out the criteria and standards of water for potable purpose.
  - b) With the help of chemical reactions, explain the procedure for the determination of hardness of water.
  - c) What is hydrologic cycle ? Explain.

(4+5+3)

- 5. a) Explain the chemical procedure used in the determination of nitrogen in waste water.
  - b) Discuss the allowed limits of radiological contaminants in water and their effect on human life when they are present in excess.
  - c) Explain the principle and gravimetric procedure for the determination of sulphate in water sample. (5+4+3)
- 6. a) What is MINAS ? Explain the MINAS specifications for sugarcane industries.
  - b) What is quality control ? Explain the function and responsibilities of quality control department in the pharmaceutical industry.
  - c) What are remote control systems ? How they are useful in chemical warfare ? (4+5+3)
- 7. a) Describe the importance and responsibilities of Quality assurance in a chemical industry.
  - b) Explain the utility of ISO 9000 system in quality control.
  - c) What is ASTM ? Illustrate the role of ASTM standards in controlling the quality with suitable example. (4+4+4)
- 8. a) What are explosives ? Bring out the procedure for the disposal of explosives in laboratory with an example.
  - b) What is the effect of leaking compressed gas cylinders? Explain the emergency response to be followed during the leaking of compressed gas cylinders.
  - c) Write a note on intellectual property rights. (4+4+4)
- 9. a) Describe the importance and procedure for verification and segregation of laboratory waste.
  - b) What are hazardous materials ? Explain the procedure for transportation and incineration of hazardous materials.
  - c) Why it is important to safeguard and secure the data properly ? Explain.

(4+4+4)